

ICP: You CAN'T Just Push Play

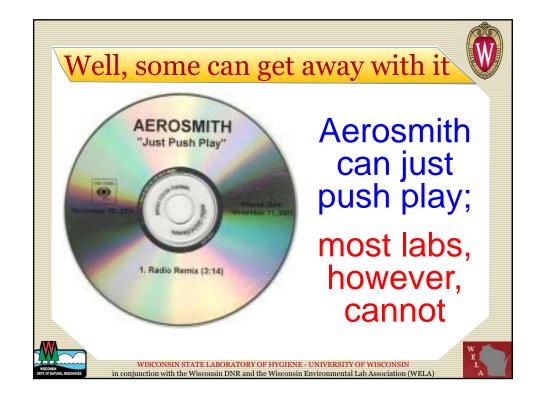


A discussion and compelling argument supporting the rationale behind and critical need for proper and effective use of inter-element correction factors (IEC) —or other interference correction tools— to obtain quality ICP-OES results on environmental samples.









Disclaimer



Any reference to product or company names does not constitute endorsement by any of the following:

- Wisconsin State Laboratory of Hygiene,
- University of Wisconsin,
- WI Dept. of Natural Resources



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Cast of Characters

State Lab of Hygiene

DeWayne Kennedy-Parker Kevin Kaufman

Roger Schultz

Brian Clary

RJ Messling

Wisconsin DNR

Rick Mealy

Special thanks to

Paul Harris and WELA for both championing and sponsoring this event as well as providing the room and refreshments.

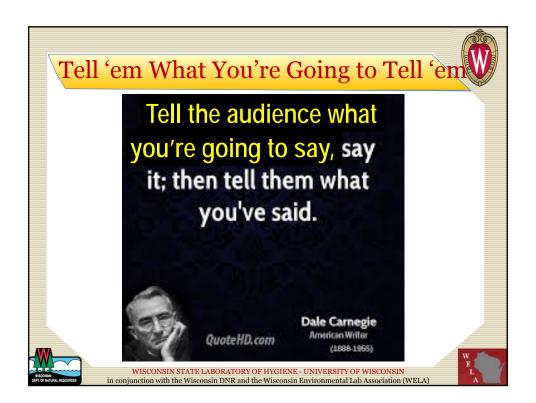


Outline



- Instruments, sample introduction, auto-samplers
- Brief instrument set-up and calibration
- Background correction (pre-loaded vs. reality)
- Inter-Element Correction factors (IECs)
- Multi-Component Spectral Fitting (MSF) and Fast Automated Curve-Fitting Technique (FACT)
- Interference check samples (ICS)
- PT sample results and examples of interference correction.
- Real World Sample Results
 Wrap-Up/ Q&A





Take home messages



- Unless your samples are ultra-trace level and matrix free, some form of interference correction is required.
- Size does not matter. Having a huge focal length is not a substitute for interference correction.
- MSF and FACT are approved forms of interference correction.
- Test your interference correction technique by appropriate design of ICS samples.
- Use appropriate evaluation criteria for ICS.

 Hint: referenced method criteria are not appropriate.
- The "CLP" style ICS-AB offers NO value.



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Instrument array (survey)



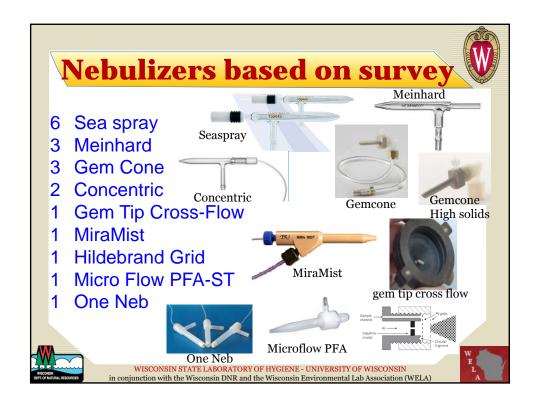
- 5 **Agilent:** (3) 720-ES, (1) 720, (1) 710-E
- 3 **J-Y:** (2) Ultima 2, (1) Ultima 2C
- 1 Leeman: Profile Plus
- 6 Perkin-Elmer: 2100, 3500, 4300,

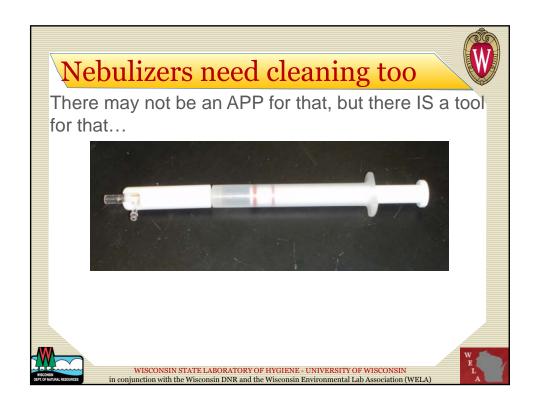
Optima: 5300, 7000, 7300

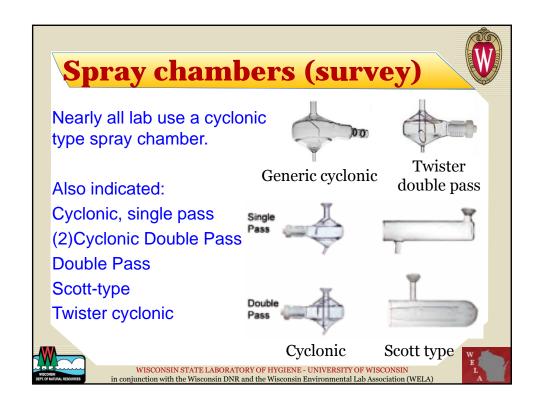
3 Thermo: (2) iCAP 6300, (1) iCAP 6500



W E L











Instrument Set-up:



- Sample (and internal standard) flow check
- Spectral alignment using a single element (Mn, etc.),
- Internal instrument alignment (Hg),
- Equipment inspection
- Documentation as required by accrediting agency





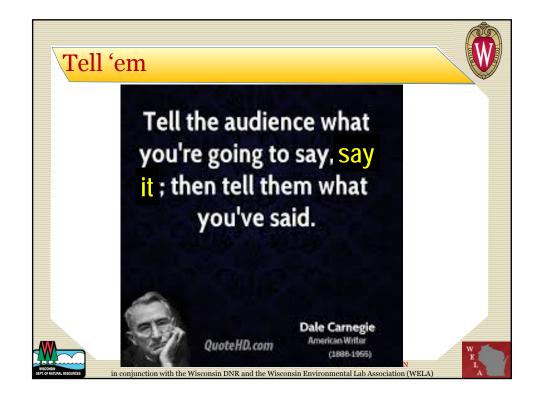
Calibration...a few words

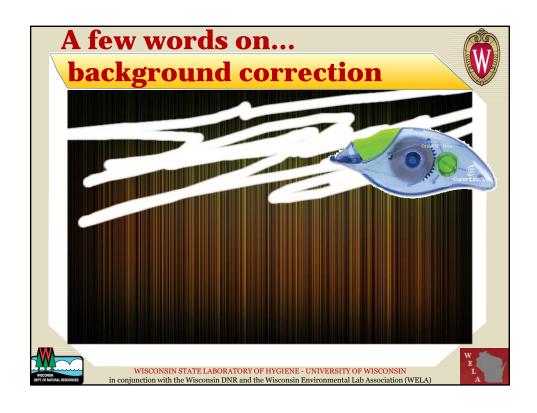


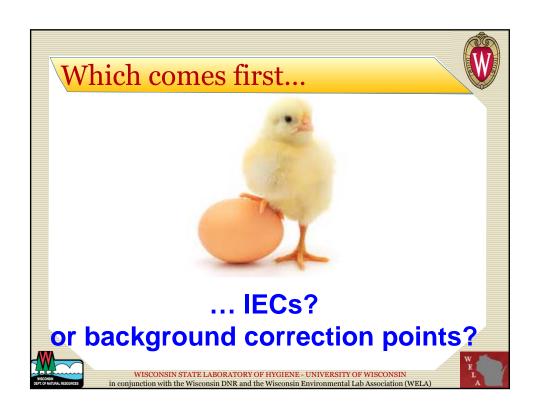
- 1pt. vs. multi pt. (ELLAP),
- element separation
- conc. levels (MDLs and linear dynamic range),
- wavelengths











It's a trick question...



You can do these steps in either order, but because IECs have to be verified if background correction points change, it makes sense to set background points first.

You can update an IEC without changing background correction pts.

But if you change background correction (location of a point, going from 1 to 2 pts or 2 to 1 pt), you MUST re-assess any associated interference correction



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It's a trick question...



Background correction points and IECs work in sync.

Proper background placement or switching from 2 to 1 pts is not necessarily a substitute for IECs.

2 background points is generally reserved for areas of sloping background.

When the sloping background is a shoulder of an interferent, you will need to have an IEC to recoup the area "shaved".





Conclusion:



You can achieve accuracy either way.

You just have to remember that IECs are inherently linked to the background corrections points designated when then IECs were generated.

Changing background correction points requires

- 1. Re-standardization
- 2. Re-evaluation of any associated IECs.
- 3. Run controls (ICS) to verify

But re-evaluating IECs does not require re-setting background correction points.



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INTERFERENCE CORRECTION





Interference Correction Rules



- 1. Not mandatory
- 2. No correction is a special case
- 3. Correction specific to configuration
- 4. Correction is subject to change





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Interference Correction Rules

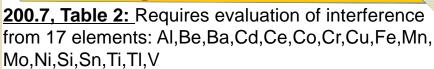


- 6. Please...no more ICS-AB
- 7. Protection is only as good as ICS
- 8. INT > ICS requires dilution or re-do IECs
- Size & resolution do not matter

 Evaluate ICS samples properly



Which single elements? Level?



<u>6010C, Table 2:</u> Requires evaluation of interference from 10 elements: Al,Ca,Cr,Cu,Fe,Mg,Mn,Ni,Ti,V

1000 ppm: Al,Ca,Fe,Mg (that seems a bit high: 250? 500 ppm?

200 ppm: all others (that may also be a bit high; 10 ppm will work in most cases)



Note that 6010 doesn't require evaluation from Mo. That wouldn't have ended well here, now would it?





Procedure to set IECs



- 2. Determine LODs.
- 3. Calibrate.
- 4. With IECs turned **OFF**, run single element standards for the 17 key interferents.
- 5. Populate an IEC table based on the results.
- Analyze appropriate ICS standards designed to test for adequate correction of these interferences.
- Verify that LODs are realistic (bouncing around the LOD).







Fast Automated Curve-fitting Technique



Minor wavelength offset or drift that may occur is accounted for by monitoring six plasma-based emissions lines across the wavelength range. This maintains the wavelength accuracy of the models for long-term use.

Unlike IEC, the known concentration of the analyte and interferent in the respective solutions is not required. The solution concentration only needs to be high enough for the signal peak to be easily distinguishable from the background (typically 50 x LOD).



Source: Agilent Technologies, Inc. 2009 Publication number: 5991-0846EN
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How does FACT work?

Mathematically "deconvolutes" (or separate) the analyte signal from the raw spectrum.

Models built by measuring the expected components separately and the response for each. This typically includes measuring these solutions:

1. A blank solution

←Here, it's

2. A pure analyte solution

much like

3. Pure interferent solutions.

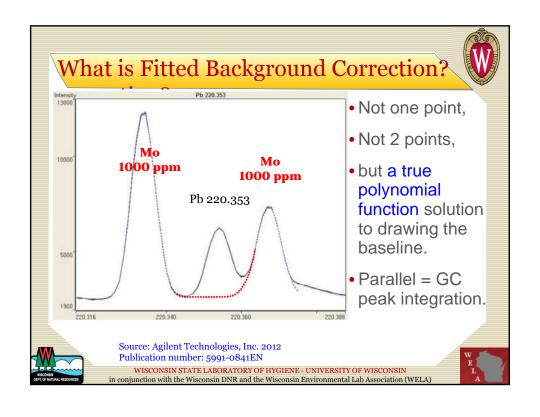
MSF

Up to 7 interference models per analyte



Source: Agilent Technologies, Inc. 2009
Publication number: 5991-0846EN
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Fitted background correction



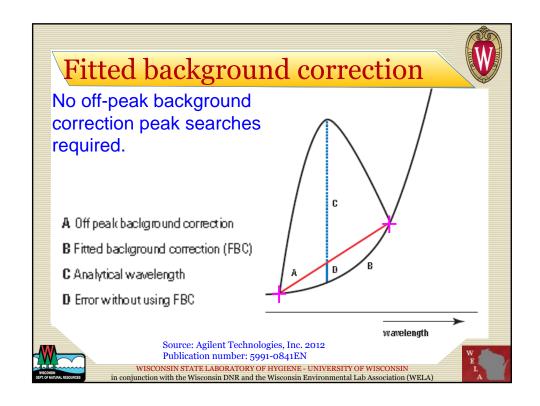
Mathematically models the measured spectrum:

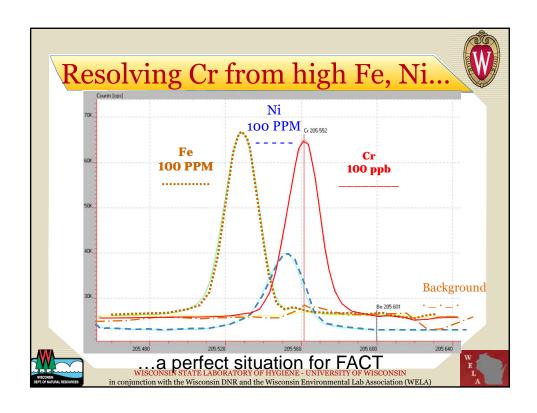
- 1. Determining the **offset component** to model the unstructured continuum background.
- 2. Determining the slope component to model the wings of large distant peaks.
- 3. Applying three Gaussian peak components to model:
 - a. The analyte peak.
 - b. Any potential interference peak left of the analyte peak.
 - c. Any potential interference peak right of the analyte peak.
- 4. Using an iterative procedure to estimate the width and position of the peaks.
- 5. Using a method of least squares to determine the magnitude of the offset, slope and peak heights.

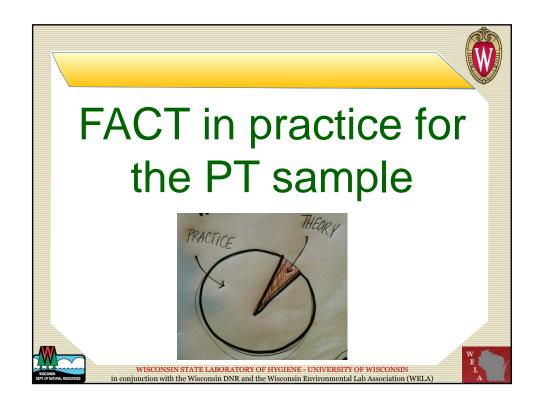


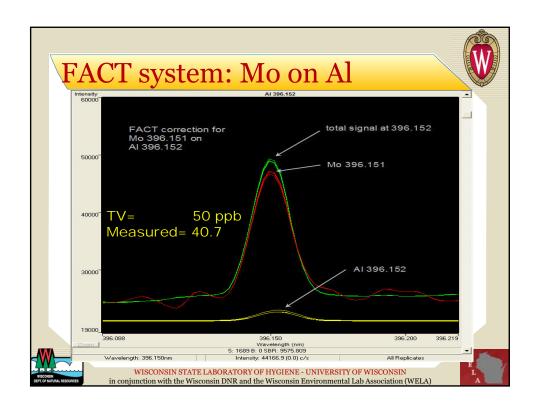
Source: Agilent Technologies, Inc. 2012. Pub. number: 5991-0841EN. 'Fitted' — Fast, accurate and fully automated background correction. WISCONSIN STATE LABORATORY OF HYGIENE - UNIVERSITY OF WISCONSIN in conjunction with the Wisconsin DNR and the Wisconsin Environmental Lab Association (WELA)

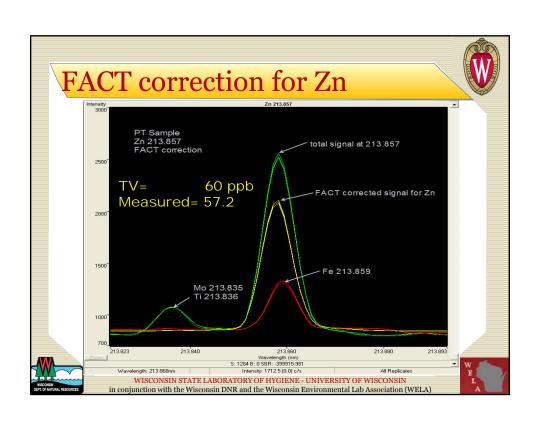


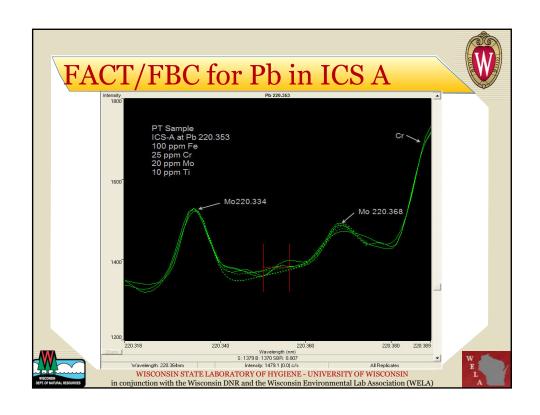


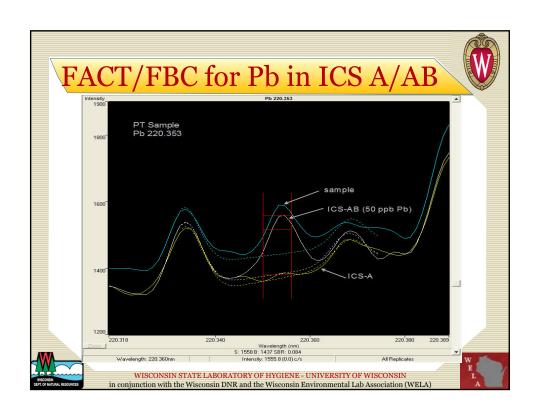


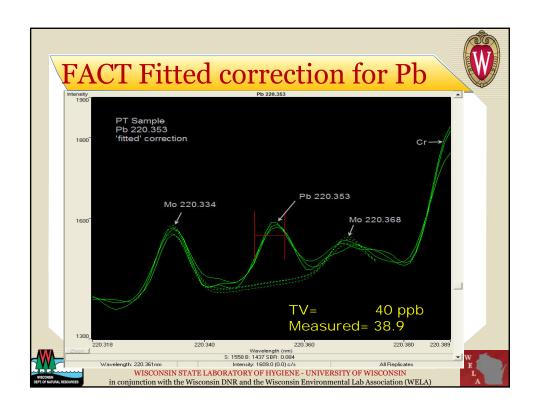


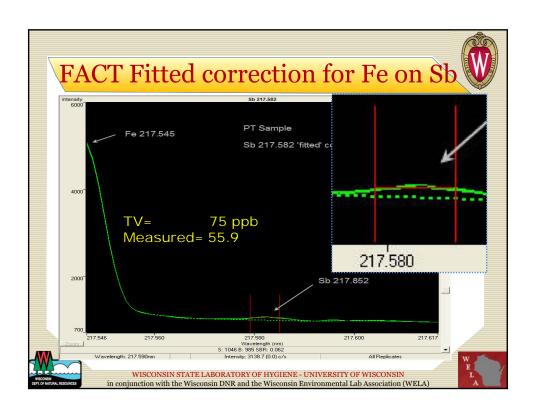


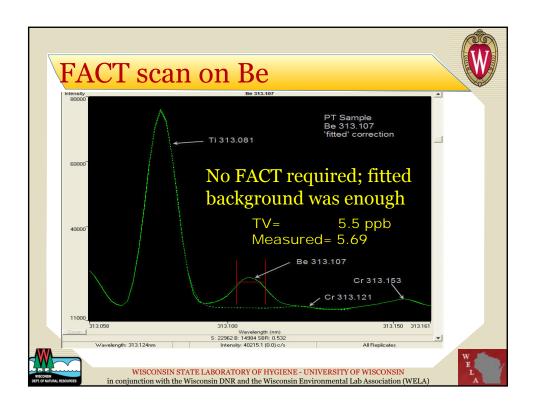


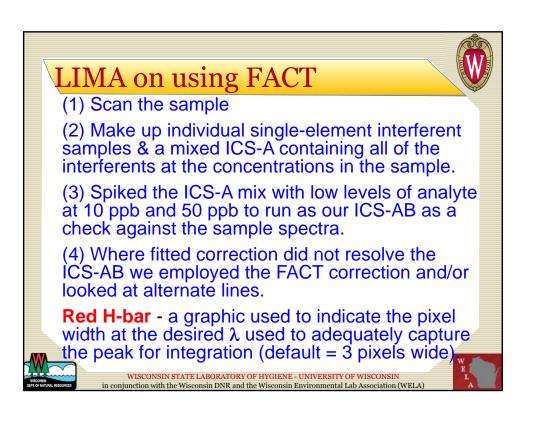




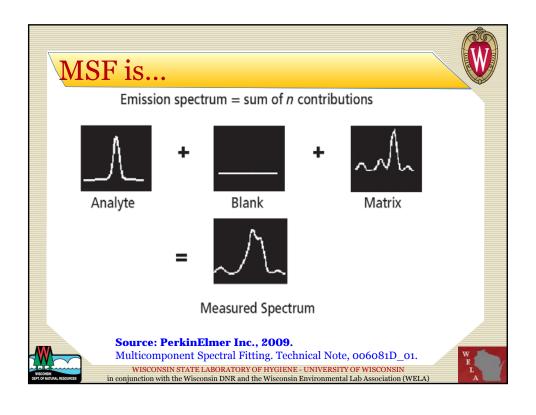












Understanding MSF



Think of it as automatic simultaneous multi-point background correction.

Requires only that a minimum of three solutions are analyzed:

- the blank,
- a pure solution of the element being determined,
- and pure solutions for each of the potentially interfering elements in the matrix.

There are no limits on the number of interfering elements that can be included in a model.



Source: PerkinElmer, Inc. 2009

Multi-component spectral fitting. Pub. # 006081D_01

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Understanding MSF



MSF looks at the spectral profile in the region around the wavelength of interest and not just at the target wavelength as is the case with IECs.

What MSF can't correct is direct spectral overlap. As long as there is some peak separation between the target element and the interference, then MSF will work.

In cases where the background interference involves multiple peaks from one element that creates "spectral grass", IECs simply do not work as background is larger than the target signal.

Bottom line is that neither IECs or MSF can handle all interferences. There are some specific interferences that can only be addressed by one vs the other.



Source: Personal communication, 2014
Perkin-Elmer Atomic Spectroscopy Product Specialist
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MSF in practice



A user dialog box allows the analyst to select whether a particular line should be considered a blank, an analyte, or an interferent.

The scan appears no differently.

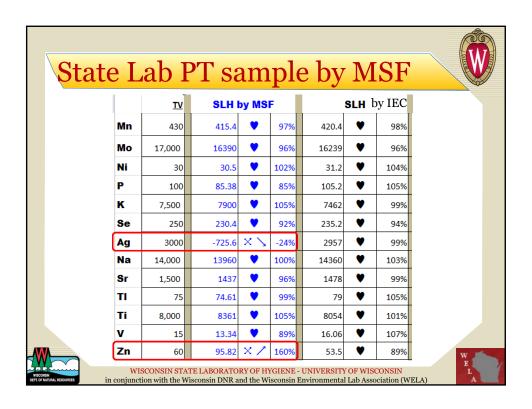
The SLH has MSF capability but generally does not use it because it's more sample specific. However, an MSF routine was quickly created for the PT sample.

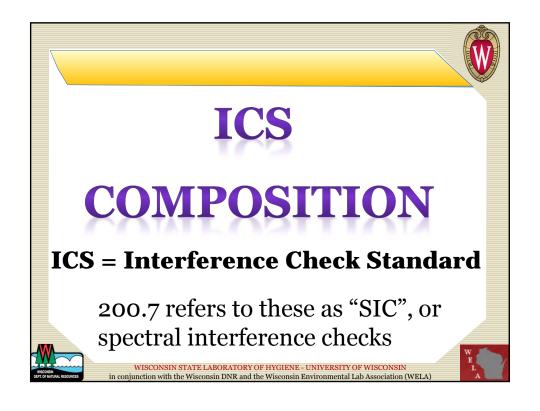


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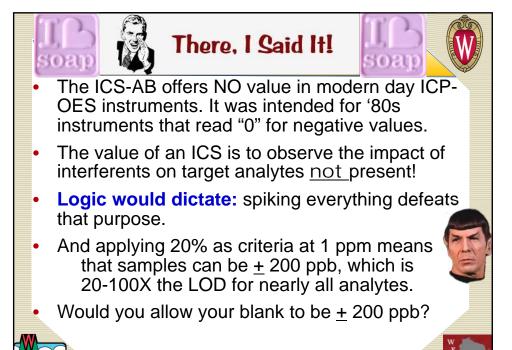


State Lab PT sample by MSF **SLH by MSF SLH** by IEC <u>TV</u> 50 75.96 • 152% ΑI 49.6 Sb 75 58.8 78% 67.4 90% 110% As 100 109.5 98 98% Ba 880 860.7 98% 900 102% 5.5 5.556 101% 5.2 В 750 679 91% 112% 842 Cd 35 33.59 96% 35.2 101% Ca 100% 13,500 13460 13292 Cr 25,000 24030 24610 98% 96% Co 240 230.6 235.7 98% Cu 1,300 1275 98% 1315 101% Fe 110,000 109400 112185 102% 99% Pb 40 42.23 106% 98% 39 Mg





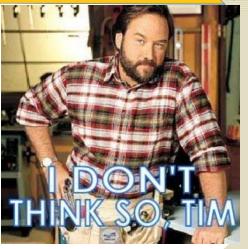








Blanks <u>+</u> 200 ppb?





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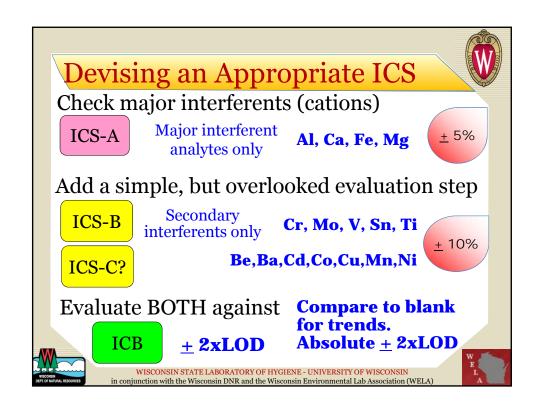
ICS Evaluation Criteria

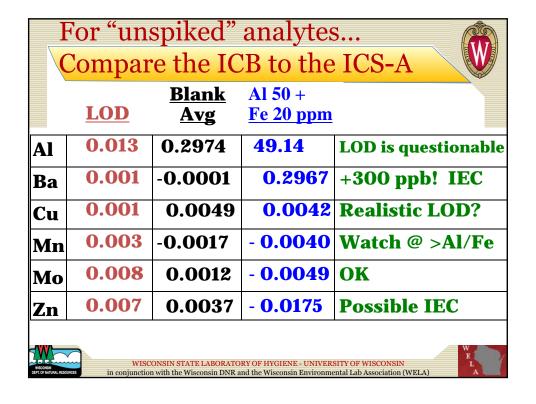


- Can we all agree that using <u>+</u> 20% as acceptance criteria for ICS samples is just plain silly?
- Yes, that's what the methods suggest, but...
- ICV criteria are ± 5% (200,7) or ± 10% (6010)
- CCV criteria are <u>+</u> 10% (200.7,6010)
- LCS criteria are ± 15% (200.7)
- At the ICS levels, shouldn't + 5% be achievable?
- And for unspiked analytes, shouldn't they be the same as in a blank?
- Using + 20%? You may as well not do it at all.









Compare the ICB to the ICS-B



	<u>LOD</u>	<u>Blank</u>	ICS-B*	
Ag	0.0005	0.0009	-0.0167	Some interference
Al	0.013	0.2974		Significant ↑ IEC!
Cd	0.001	-0.0021	0.0003	Looks OK
Zn	0.007	0.0037	-0.0139	Small IEC needed?

*Cr,Cu,Mn,Ni,Ti,V @10ppm



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PT Sample Results In 2005, we presented The "Wibby"... In 2014, we upped the ante Wisconsin State Laboratory of Hygiene - University of Wisconsin in conjunction with the Wisconsin Environmental Lab Association (WELA)

PT Sample Results

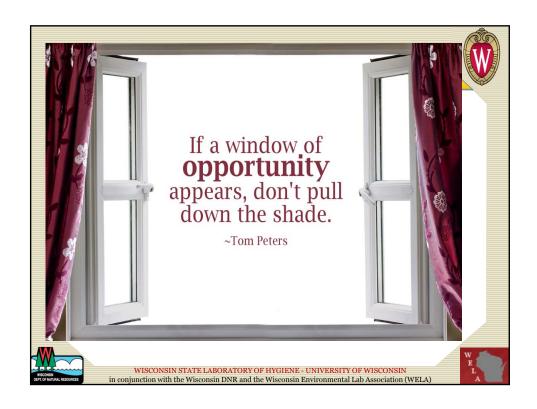


Remember...the point of these samples was analogous to an antiviral application for a computer; the sample was designed to highlight potential gaps in your interference security system. If your lab did not fare well, please do not be discouraged, we will be spending the majority of our time highlighting what the likely causes of these results are and how to correct them.

Please also understand that these results are in no way reflective of any lab's day-to-day performance using ICP. These samples, particularly the home brewed one, were specifically presented because of the challenge they pose in terms of interferents and actual levels of interferents.

Most labs will not encounter samples with such severe interferences routinely, but in the event you do, this will help you come up with a plan to upgrade your defense systems.





PT Sample Game Plan



Interferences:

Fe (110 ppm) Cr (25 ppm), Mo (17 ppm), Ti (8 ppm)

LOD challenges:

AI (50 ppb), Sb (75 ppb), Be (5.5 ppb), Pb (40 ppb),

V (15 ppb)

Contaminant issues:

Zn (60 ppb)

Precipitation (?) loss:

Ag (3 ppm)

Every lab analyzing them did fine on the following elements: Ba, Ca, Cr, Cu, Fe, Mg,

Mo, Na, Sr, Ti



10 out of 27 elements passed by all

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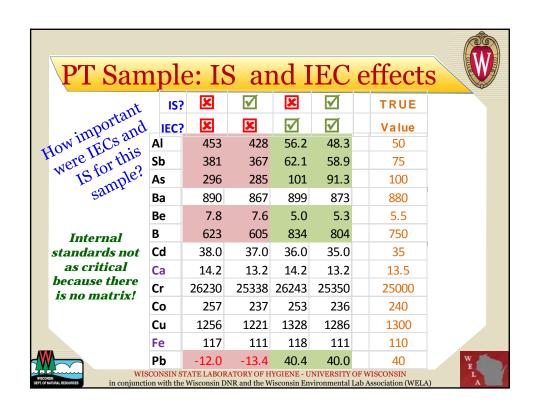
...but there was trouble w/these

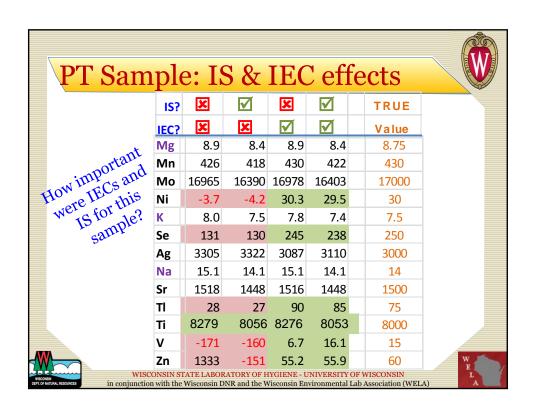


	#repto	l #ok	↑ bias	♦ Bias	False -
V	11	2	3	2	4
SI	13	5	7	0	1
Be	e 14	6	0	5	3
A	14	7	6	0	1
P	7	4	3	0	0
Zı	1 17	10	7	0	0
T	13	8	1	2	2
B	12	9	2	1	0
Pl	17	13	0	3	1

WISCONSIN DEPT: OF NATURAL RESOURCES



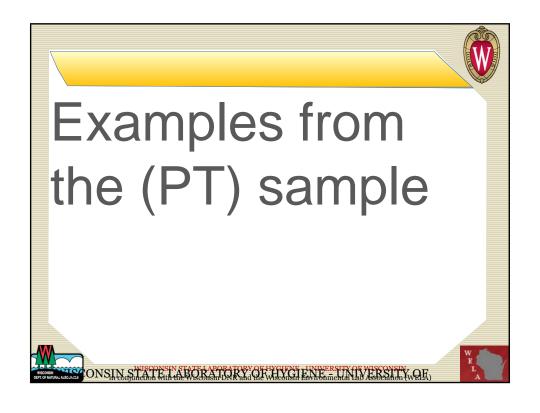


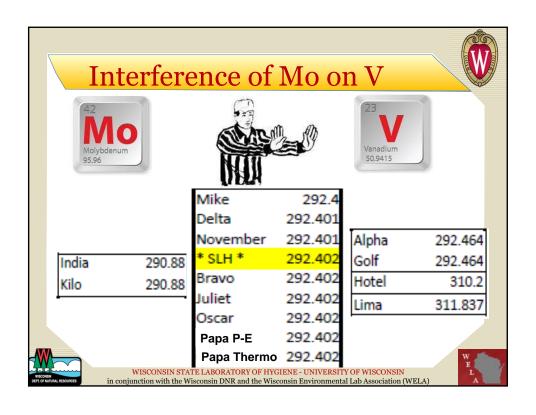


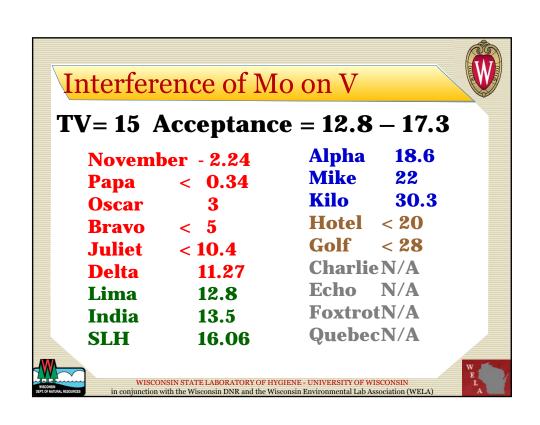
PT sam		All axial		Ti IEC	
			-Mo,Ti IEC		
Al 396.153	50	490	490	49.6	Mo
B 249.677	750	764	758	842	other
Be 313.107	5.5	5.1	13.6	5.2	Ti
P 178.221	100	61.8	57.8	105	Mo,?
Pb220.353	40	- 13.7	- 16.2	39.0	Mo
Sb206.836	75	- 144	- 122	67.4	Mo,Ti
Se196.026	250	235	235	235	Neither
TI 190.801	75	79.0	2.0	79.0	Ti
V 292.402	15	- 7.2	- 0.9	16.1	Mo,Ti
Zn 206.200	60	46.3	46.3	53.5	Mo
were m	ajor prob	olems fo	or difficu	nt Mo and lt analyte FOF WISCONSIN I Lab Association (WELL	SWEL

						n Ti	11	
			Before	After		Before	After	
			Mo 202.031		% change	Ti 336	5.121	% change
			2 ppm	17 ppm		2 ppm	8 ppm	
N S	396.153	axial	24.7235	<u>26.9535</u>	109%			
b 2	206.836	axial	-14.0392	<u>-12.935</u>	92%		2.755	
ls 1	188.979	axial	5.789	7.2234	125%	-1.97	<u>-3.7829</u>	192%
Ba 2	233.527	axial	-0.726	-0.726	no change			
Be [313.107	axial		-0.00583	new	0.85583	1.06093	124%
3 2	249.677	axial	-1.9665	<u>-4.79047</u>	244%			
d 2	214.440	axial						
a	317.933	radial						
r 2	205.560	axial	0.863444	0.51334	59%			
o 2	228.616	axial	-1.62173	-1.62173	no change	2.34087	2.34087	no change
u (327.393	axial				-6	<u>-6.0604</u>	101%
e 2	238.204	radial						
b	220.353	axial	-2.22817	-3.22628	145%		-0.3156	new

	Changes in IECs from 2 ppm to 17 ppm Mo, 8 ppm Ti									
			Before	After			Before	After		B
			Mo 202.031		% change		Ti 336.121		% change	3
			2 ppm	17 ppm			2 ppm	8 ppm		
Νg	279.077	radial								
VIn	257.610	axial								
VIO	202.031	axial						<u>0.63785</u>	new	
Ni	231.604	axial	-2.43243	-2.19289	90%					
•	178.221	axial		<u>-2.66056</u>	new			<u>-0.4928</u>	new	
K	766.490	radial								
Se	196.026	axial								
٩g	338.289	axial					-6.045	<u>-5.8001</u>	96%	
Na	589.592	radial								
Sr	407.771	radial								
П	190.801	axial					-11.6478	<u>-9.5717</u>	82%	
Гі	336.121	axial		<u>0.16409</u>	new					
/	292.402	axial		-1.4245			0.91222	<u>0.77957</u>	85%	
Zn	206.200	axial		<u>-0.43591</u>	new					
TI Ti V Zn	336.121 292.402	axial axial		-1.4245						







Interference of Mo on V

P-E default background correction points are at - 27 and +27 pm off line from the peak maximum at 292.402 nm.

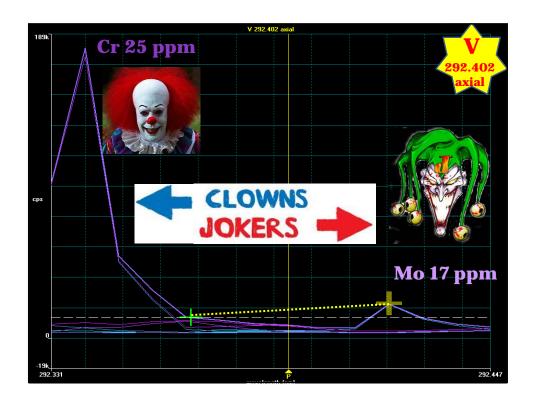
The +27 pt falls dead on top of a Mo peak (which would only be a problem at these Mo levels).

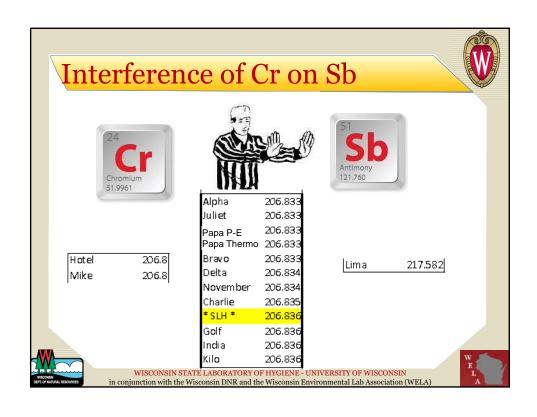
Kilo (202%) uses 290.88 line (P-E 1st recommended line). There is also a Mo peak at 290.912.

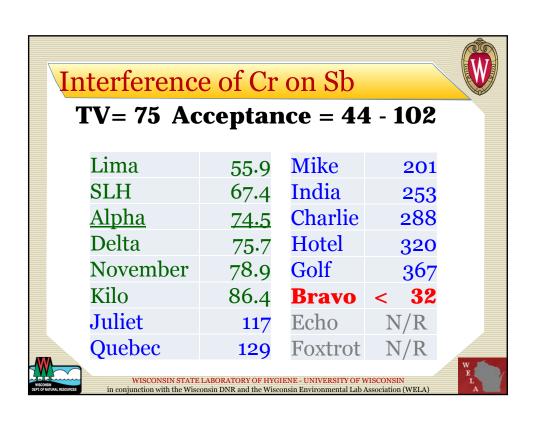
Lima passed but had low bias. They use λ 311.837 (5th recommended line). There is a Cr line at 311.865 nm.











Interference of Cr on Sb

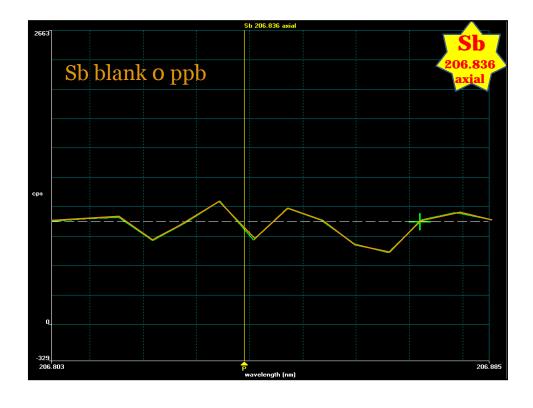


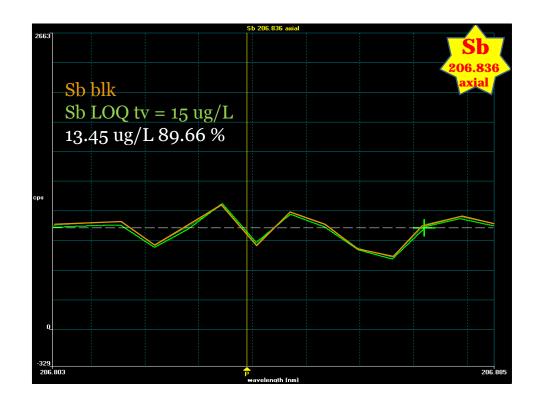
High bias will result (676% for SLH) if no correction was employed for interference due to Cr.

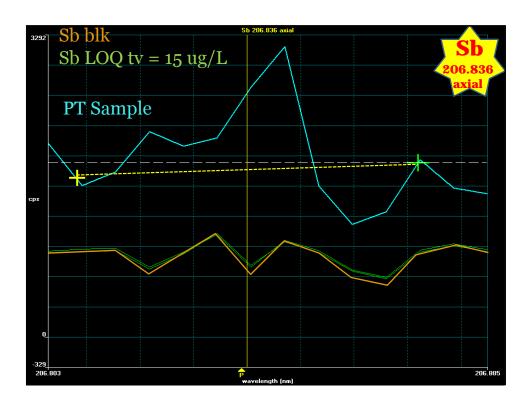
False "-" is most likely due to use of 2 background correction points. Cr is right on top of Sb.

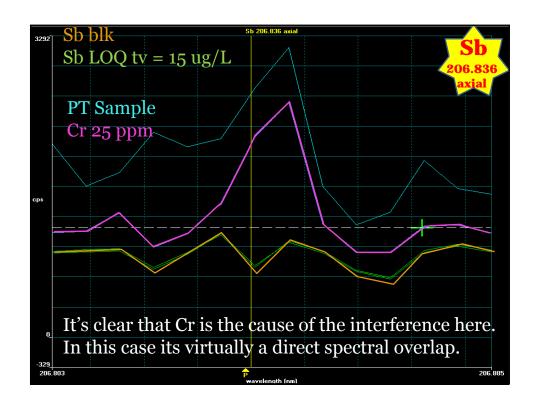


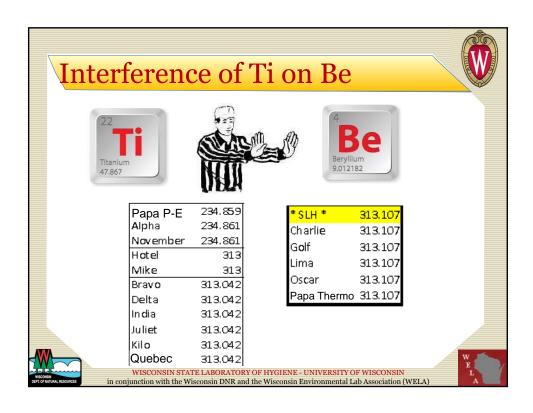












Interference of Ti on Be



Charlie < 0.79	Quebec	5
Hotel < 1	Delta	5.069
Kilo < 1.0	India	5.14
Papa 1.7	Juliet	5.19
Oscar 4	SLH	5.2
November 4.38	Bravo	5.26
Mike 4.4	Lima	5.69
Alpha 4.6	Golf	< 7



Echo Foxtrot N/R
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conjunction with the Wisconsin DNR and the Wisconsin Environmental Lab Association (WELA)



Interference of Ti on Be



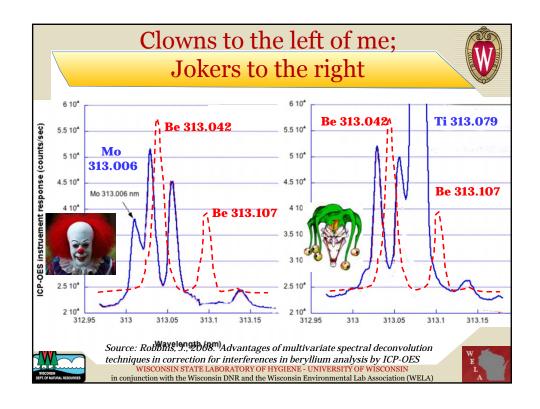
Due to Ti, even if using only 1 background correct point, low bias can result. Mo can also be an issue.

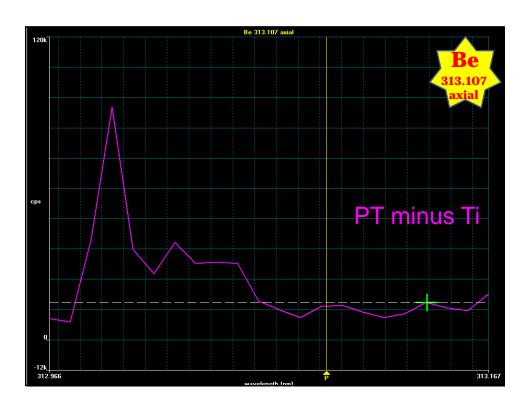
Will yield slight negative result if not using an IEC (or some other form of correction).

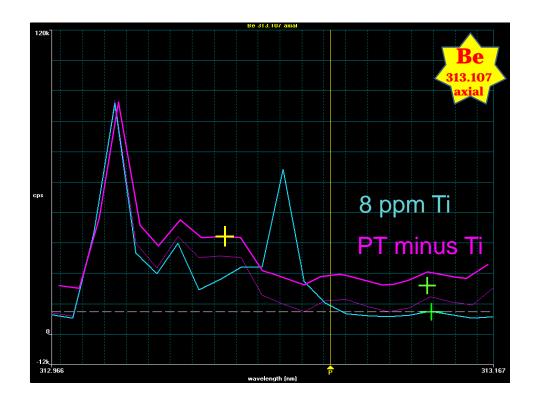
The false negative (Kilo) is suspected to result from using the lower wavelength. The background correction point (or one if using two points) is affected by neighboring Mo peak

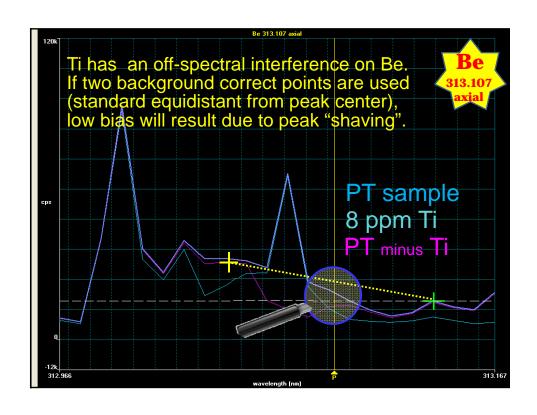


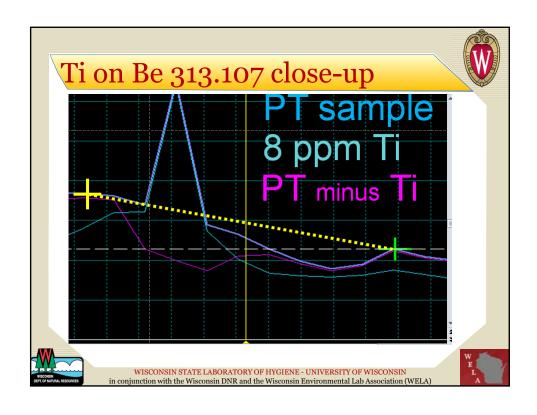


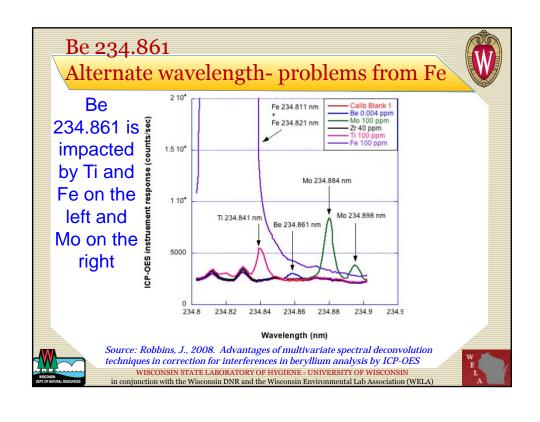


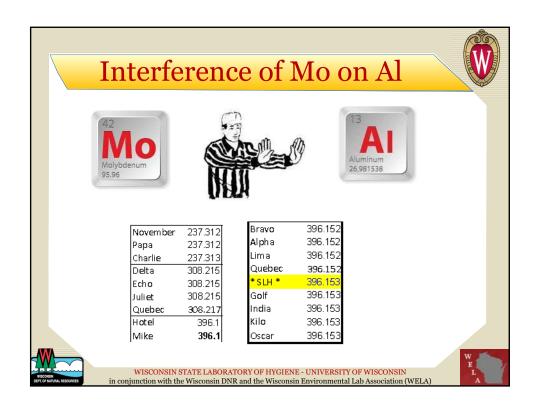


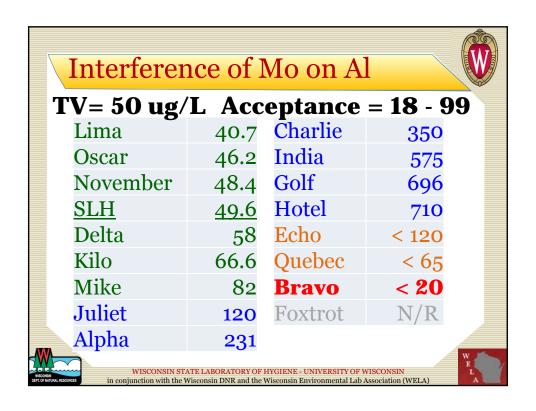












Interference of Mo on Al



Problems with AI on the PT sample were purely related to interference from Mo.

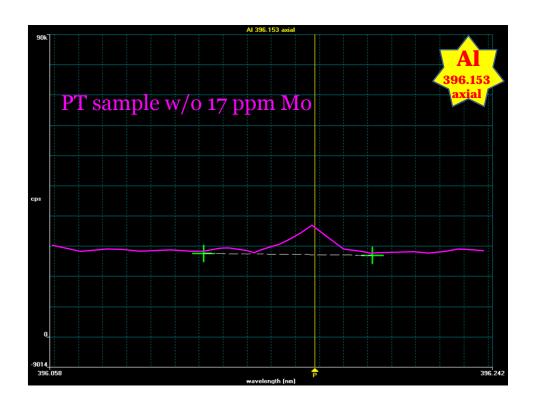
At 50 ppb, the level AI was somewhat of a challenge, but well above the mean/median LODs reported.

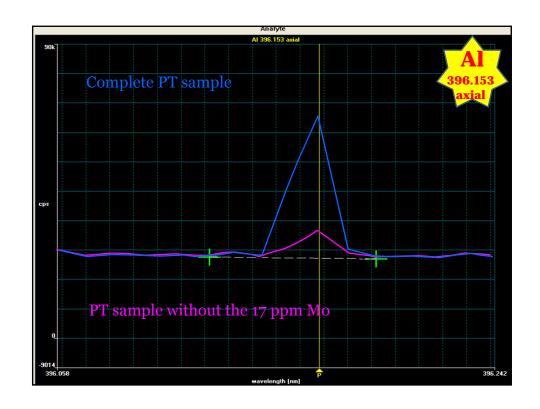
High bias observed is related to either not having interference correction for Mo, or having set the correction at a much lower level than that of Mo (17 ppm) in the PT sample.

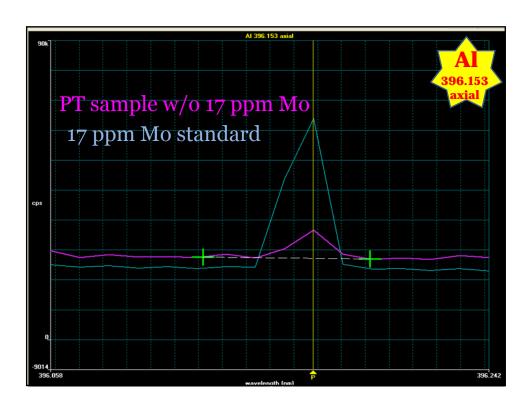
We suspect the false negative to be a result of either an unrealistic LOD or over-correction.

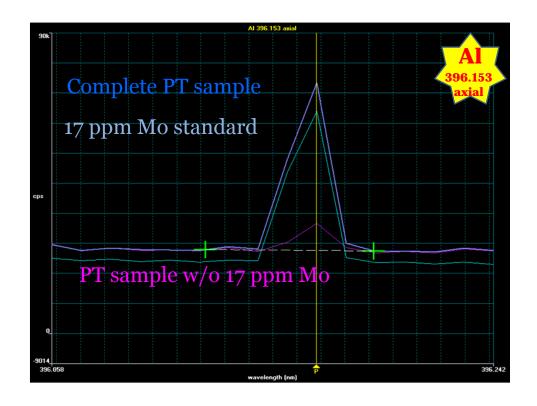


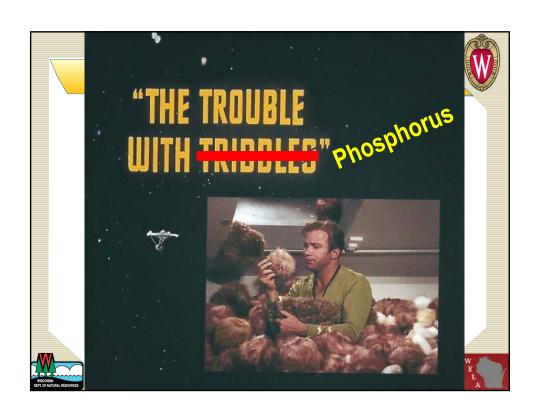


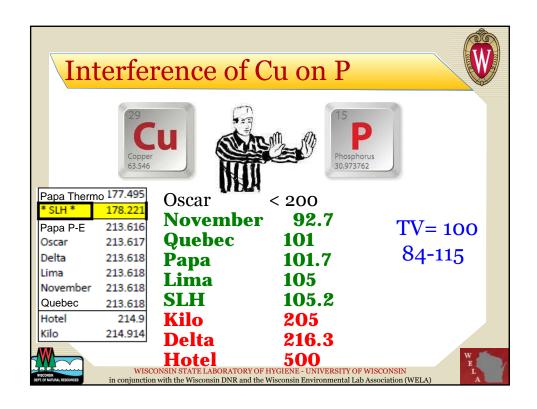












The trouble with Phosphorus



The SLH uses the 178.221 line because 213.618 and 214.914 both suffer from significant interference from Cu.

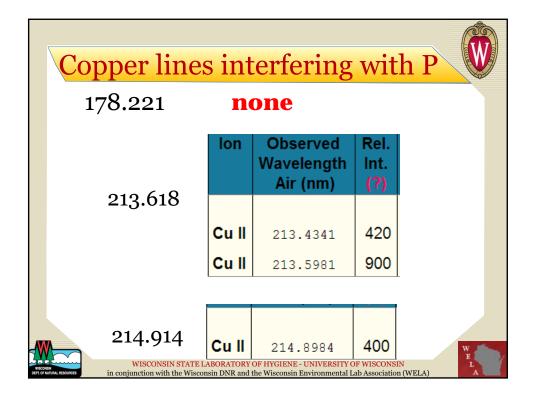
This also partially explains why the Dept. has not been keen on the use of ICP for phosphorus.

While only 8 labs supplied results, 3 of the eight (37%) succumbed to high bias (and the sample contained 1.3 ppm Cu).

The good news is that the interference CAN be controlled.







The trouble with Phosphorus

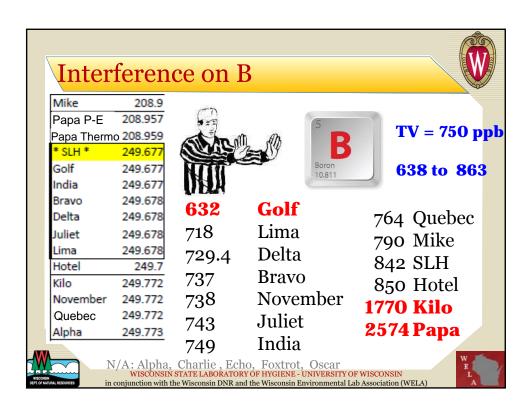


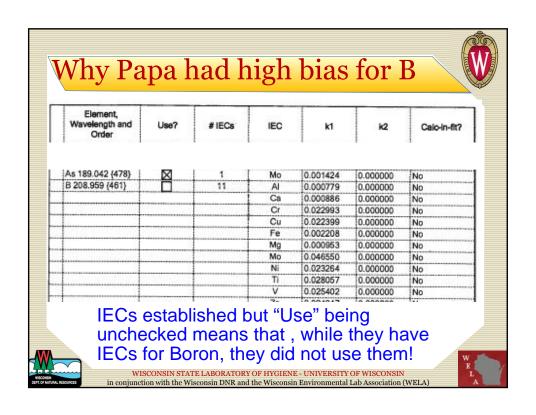
Notably, of the three labs that had trouble with P, all of them used one of the problem lines. Delta uses the 213 line and Hotel and Kilo use the 214 line.

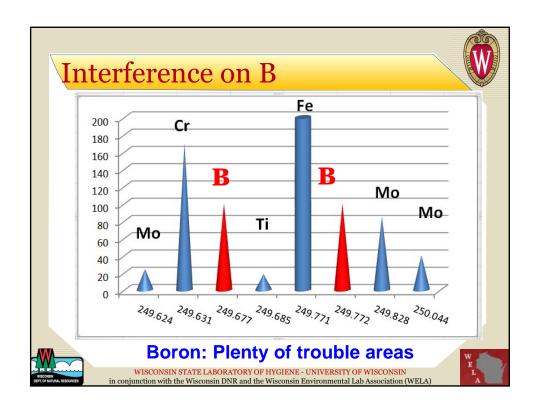
While the 178 line is less sensitive, the SLH is able to obtain an LOD of 20 ppb which, while perhaps a little high for Adaptive Management needs, is well within LOD needs for routine P in wastewater.

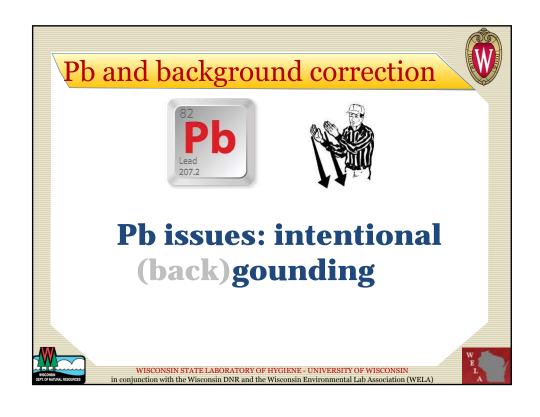


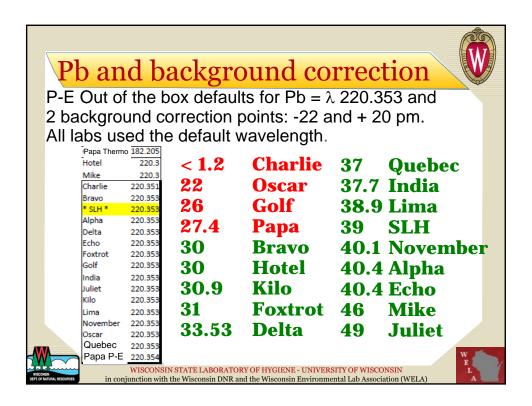
W E L

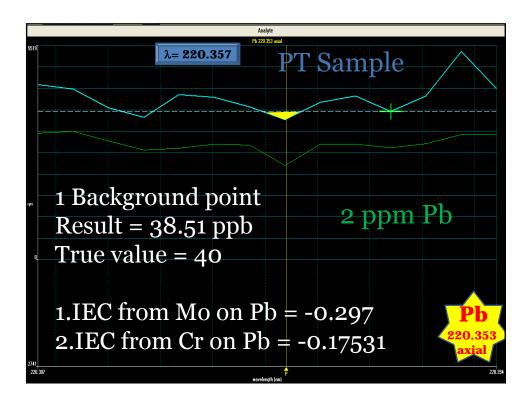


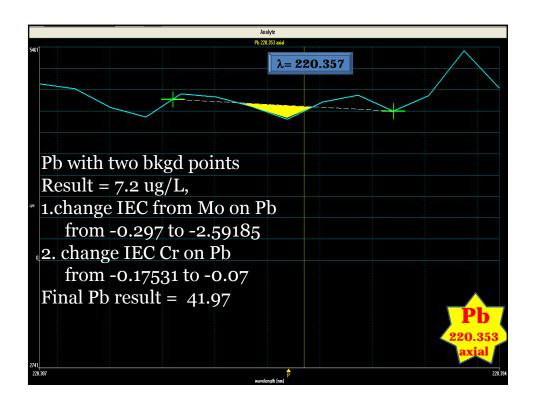


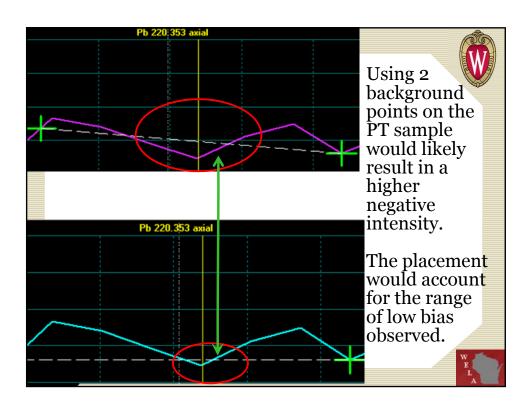












Special K





Perfect example of a background correction rule



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How many background correction points?



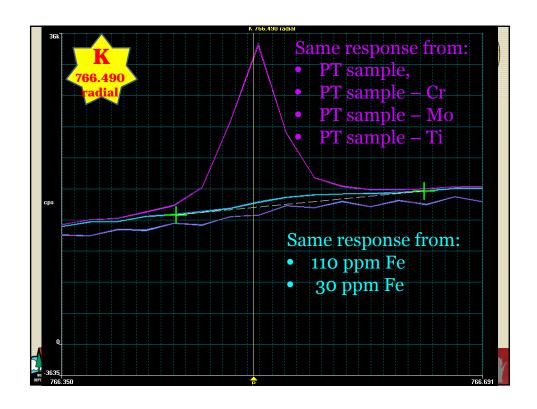
When is using 2 background correction points a requirement?

Whenever the background is sloping, you MUST use two background correction points (at least on a P-E system).

Potassium, at the upper end of the spectrum is a perfect example







PT Sample: Ag



Several labs expressed concern about the "trick" regarding analysis of silver.

Only two labs had trouble with the silver (3 ppm). Both reported results around 400-450 ppb.

One of the two reported consecutive results of : 2171, 432, and 241 ppb. All were low, but clearly silver was dropping...why?



W E L

Silver and Chloride



 $Ag^{+}(aq) + Cl^{-}(aq) \rightarrow AgCl(s)$ (precipitate)



Many analysts experience low Ag recoveries when working in HNO₃ media. The problem is due to trace chloride contamination.

But....I don't see a precipitate!

Ag has already **photo-reduced** onto the container walls. Silver is very light sensitive. You won't see a precipitate, even though it has "fallen" out.



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Handling Ag



If Ag is prepared in just a 0.5% HNO3 solution, it will not be stable for very long.

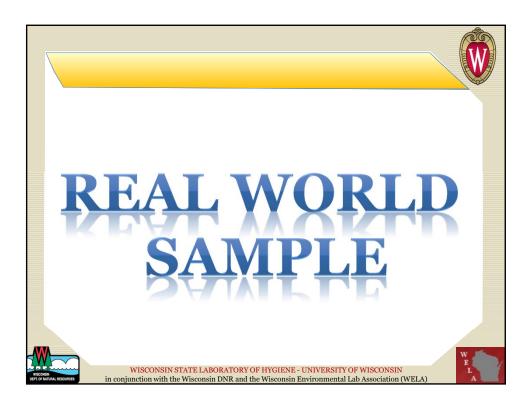
The SLH prepares Ag standard fresh daily.

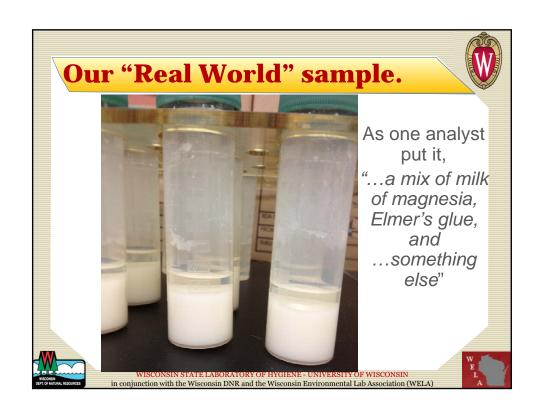
Under about 2 ppm, the presence of a small amount of HCI will actually keep the silver in solution.

We chose the 3 ppm level for Ag because at that concentration, any HCL becomes a problem and the Ag falls out of solution.

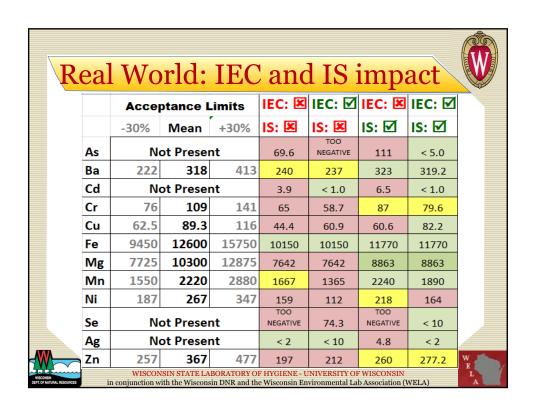
Rinse solution should have ~5% HCL (even though standards and samples are HNO3 only) to keep Ag from plating onto walls of the spray chamber











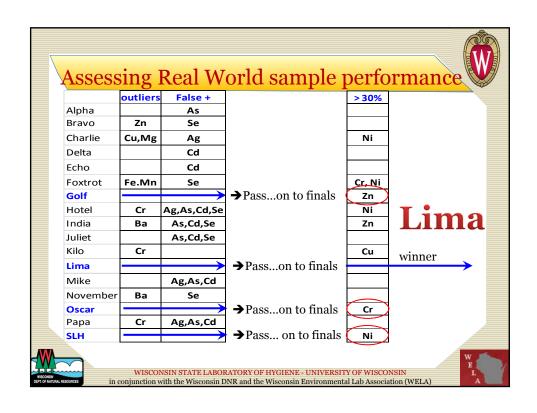
Real World Sample Concerns



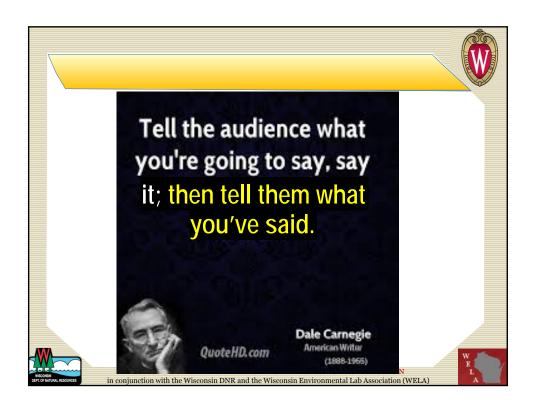
- False positives. We determined that As, Cd, Ag, and Se were not present in the sample.
- Outliers. We tested the data set for each element for outliers and, after identifying them, we removed them from the data set.
- Next we calculated the mean of the remaining data.
- We set acceptance criteria at +/- 30% of the mean.
- Any values the fell outside of 30% were marked.











This is what we told you...



- Unless your samples are ultra-trace level and matrix free, some form of interference correction is required.
- The PT results clearly demonstrate that fact.
- Establish IECs or use FACT/MSF techniques
- The Real World sample demonstrates the importance on internal standards in the face of a matrix.
- Test your interference correction technique by appropriate design of ICS samples.
- Analyzing a "CLP" style ICS-AB offers NO value.



