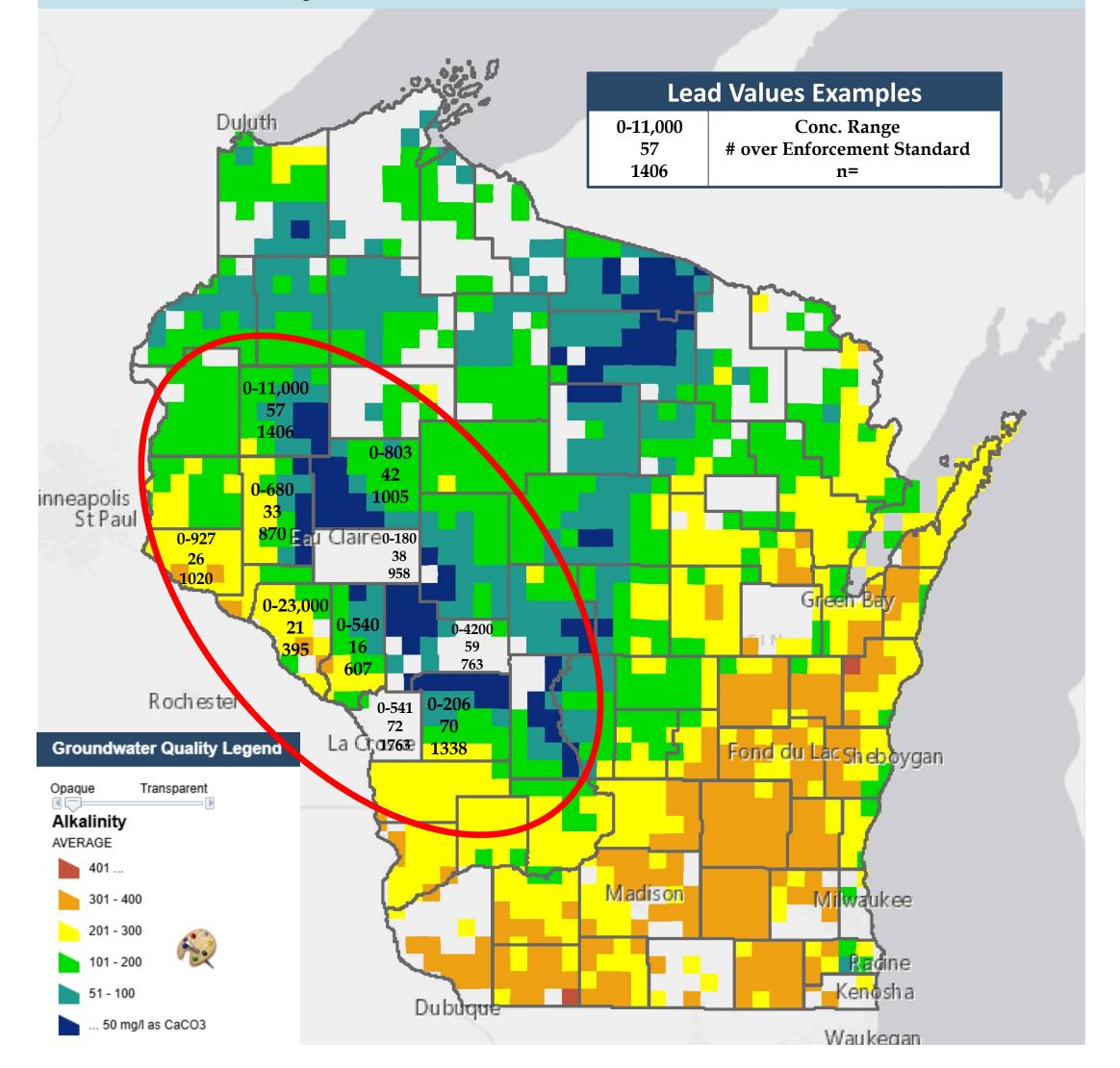
Metals in Western Wisconsin Aquifers: A Topic of Emerging Concern **Dave Johnson and Ian Anderson**

Alkalinity and Lead in Groundwater



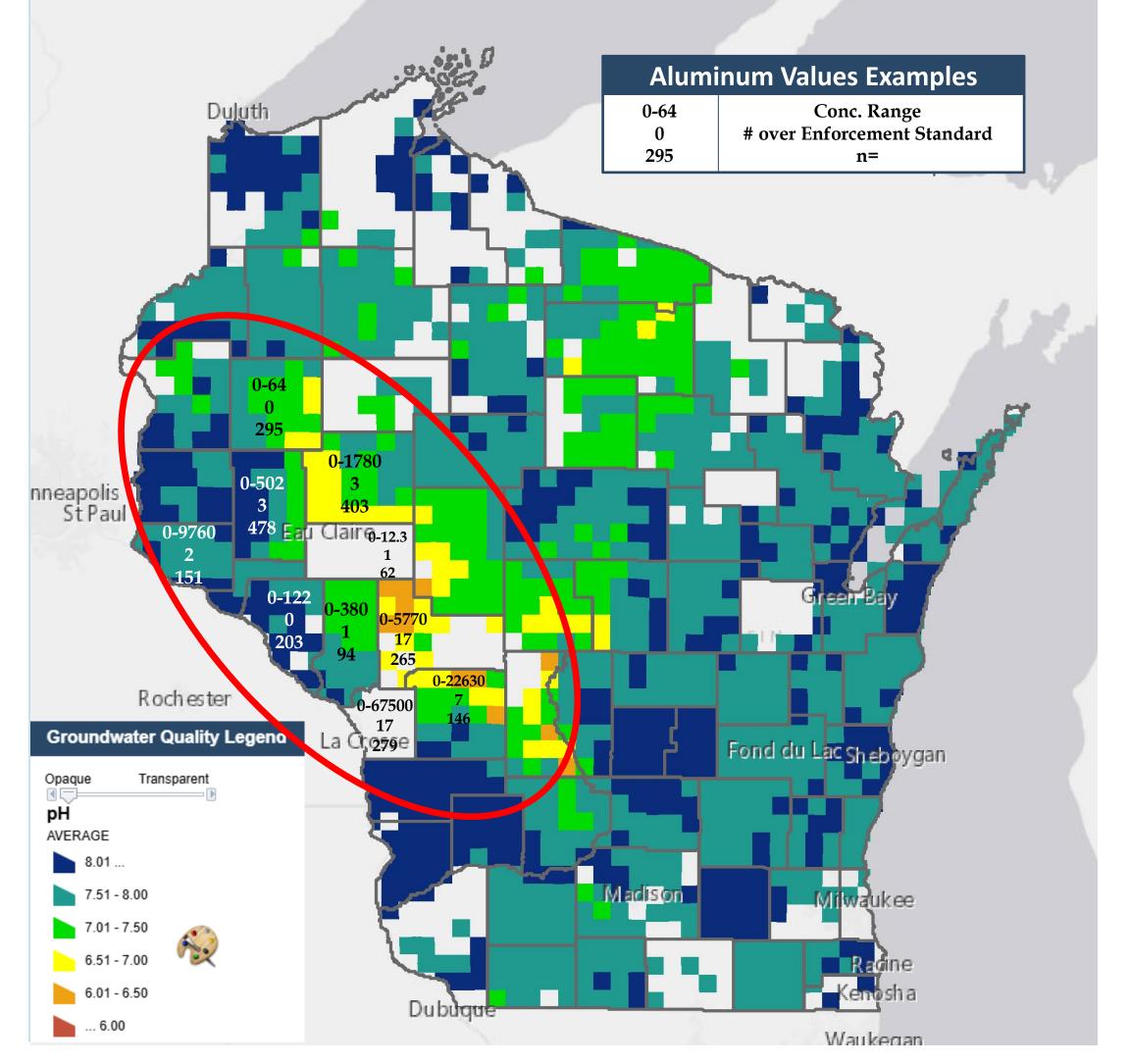
Municipality	Tomah	Bangor	Coon Valley	Stoddard	La Crosse	La Crosse		A Granddad Bluf
WUWN	ST009	UN859	OH527	CQ196	RV223	GC657	Advisory Level	La Crosse
pН	4.6	6.97	4.7	4.5*	4.08	6.85		
Sulfate	1478.5	-	-	-	-	-	250 mg/L	4m-
Aluminum	22630	3220	23900	67300	23200	2320	200 ug/L	
Arsenic	ND	ND	129	92	38	ND	10 ug/L	3m-
Cadmium	ND	2	1	5	5	ND	5 ug/L	~
Chromium	2.5	ND	77	220	56	1	100 ug/L	2m-
Cobalt	244.2	306	501	909	501	23	40 ug/L	1m-
Copper	307.5	1610	324	1290	511	16	1300 ug/L	
Iron	39.19	53.6	218	672	376	24.3	0.3 mg/L	0m-
Lead	-	ND	109	80	47	ND	15 ug/L	<u>-25</u> 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
Manganese	222.8	142	921	2980	897	56	50 ug/L	-1m-
Molybdenum	ND	-	20	-	ND	ND	40 ug/L	
Nickel	1232	419	771	1750	832	41	100 ug/L	-2m
Strontium	843.1	70	144	217	219	54	4000 ug/L	
Vanadium	ND	5	69	239	56	1	30 ug/L	-3m-
Zinc	574.6	385	52	4460	706	56	2000 ug/L	-4m-

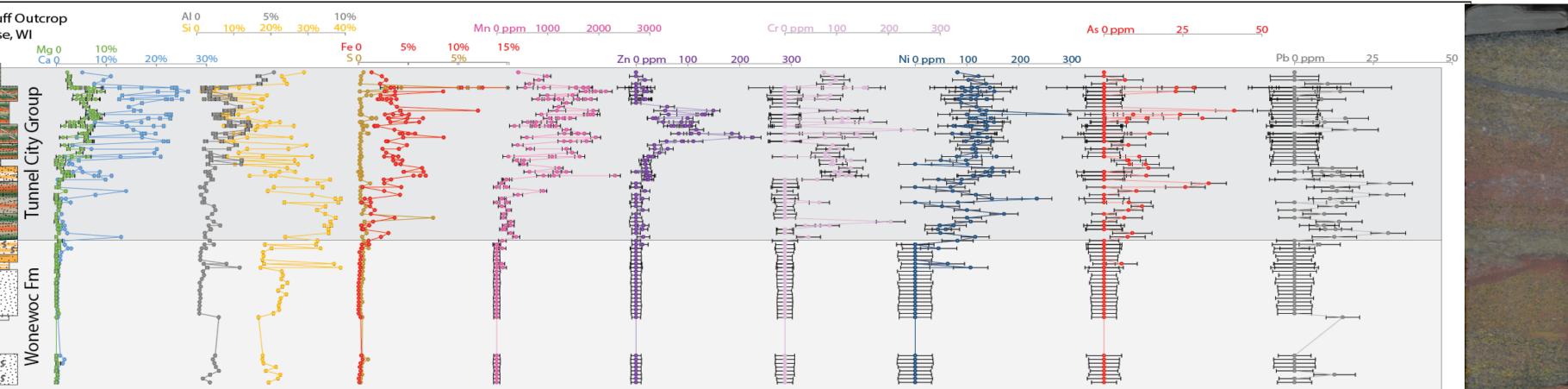
Factors That Are Currently Poorly Constrained

- Spatial Extent Discrete or Continuous?
- Mineral phase Dissolved or Activities – Which Contribute to Sorbed? Mobilization?

Note: All Data are preliminary and subject to revision

pH and Aluminum in Groundwater



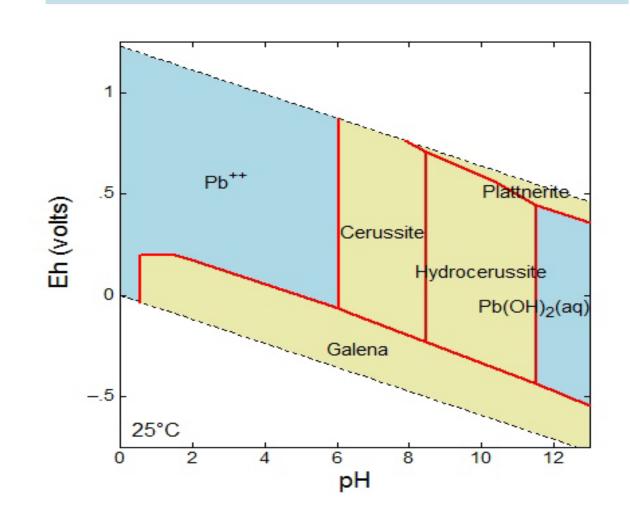


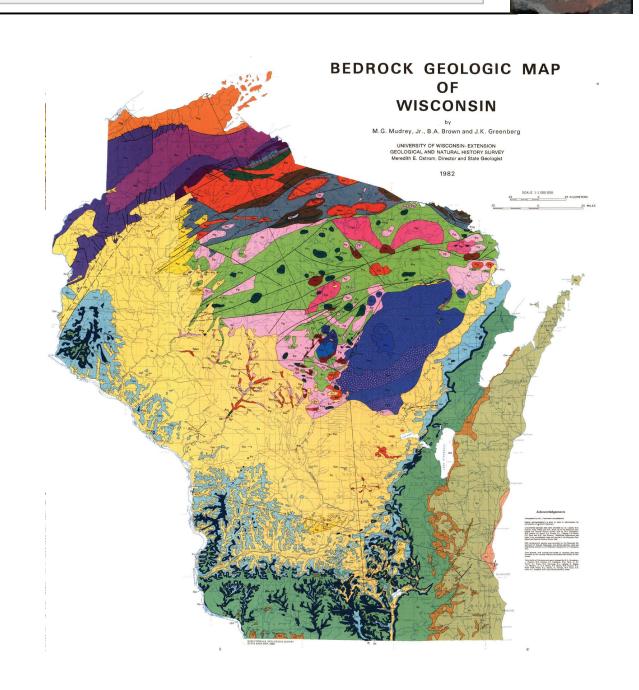
• Mineralization – Primary or Secondary?

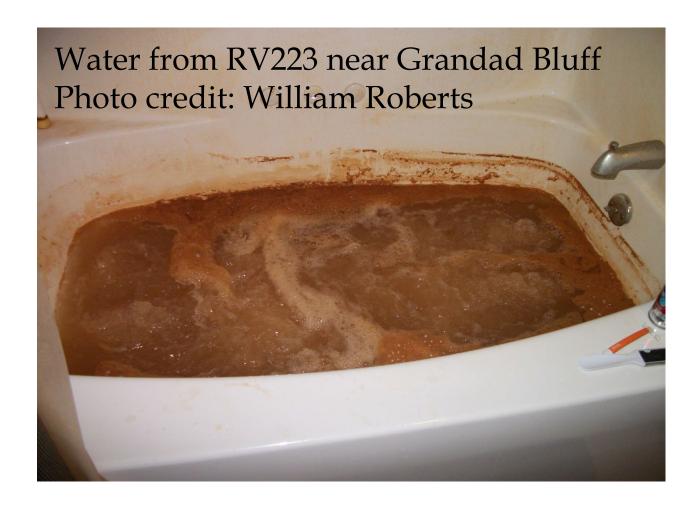


Geochemical Factors Affecting Solubility & Mobility

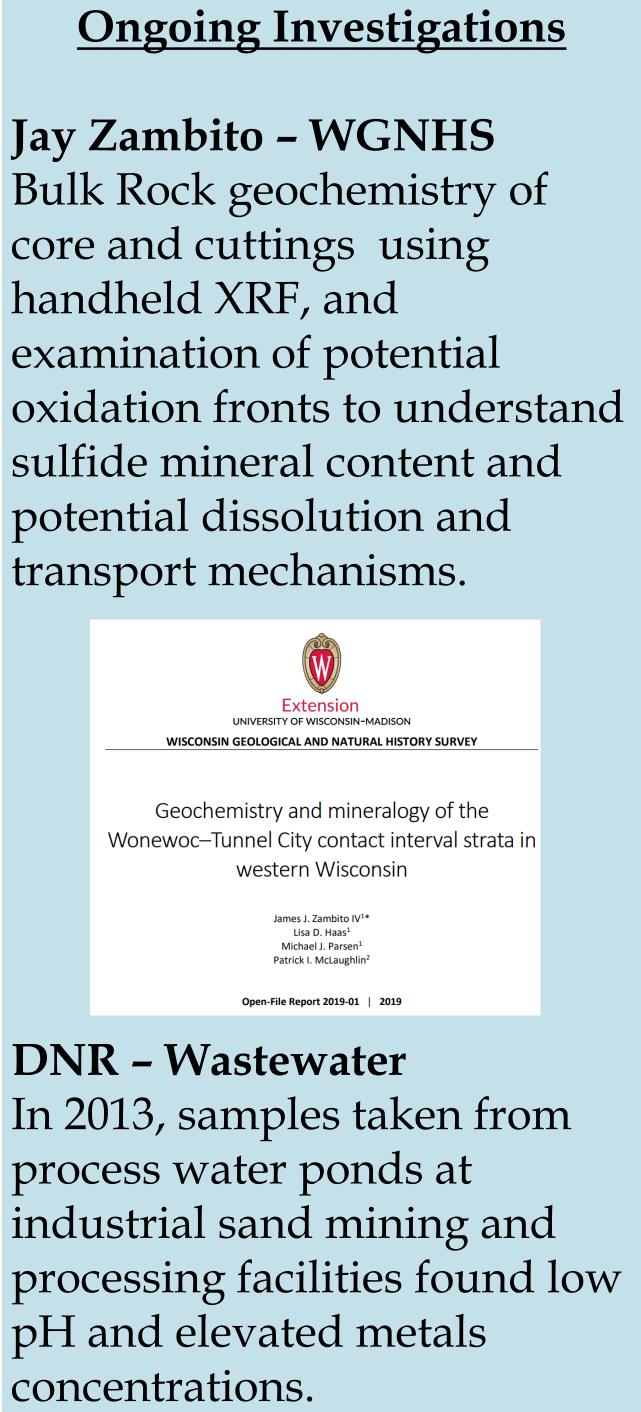
- pH
- Alkalinity
- Redox
- Mineral phase







Jay Zambito – WGNHS core and cuttings using handheld XRF, and examination of potential potential dissolution and transport mechanisms.



DNR – Wastewater process water ponds at pH and elevated metals concentrations.

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